

# **Disposable, Digital Phosphate Sensor** (OTT ID 1513)

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### Increased levels of phosphates can lead to;

- Eutrophication (excessive plant growth such as algal blooms)
- At first plant growth may be stimulated, but over time excessive plant growth can choke the water way, and lead to death of the plants
- Low oxygen in waterways can occur and death of aquatic organisms
- Some algal blooms are toxic to humans

## Sources of increased phosphates in the environment include;

- Fertilizers and farm water run-off (and manufacturing of fertilizers)
- Sewage
- Pulp and paper industry
- Detergents
- Vegetable and fruit processing





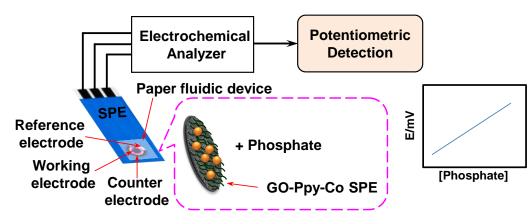
- The EPA notes that "Monitoring phosphorus is challenging because it involves measuring very low concentrations down to 0.01 mg/L or even lower...."
- Even such very low concentrations of phosphorus can have a dramatic impact on streams. Less sensitive methods should be used only to identify serious problem areas
- The EPA approved method for measuring phosphates involves the use of chemical reagents
- In some cases samples must be brought back to the lab for analysis



## **Technology Solution**

## Low Cost, Disposable Phosphate Sensor

### Schematic diagram of the setup



### Chemical reaction using Cobalt (example)

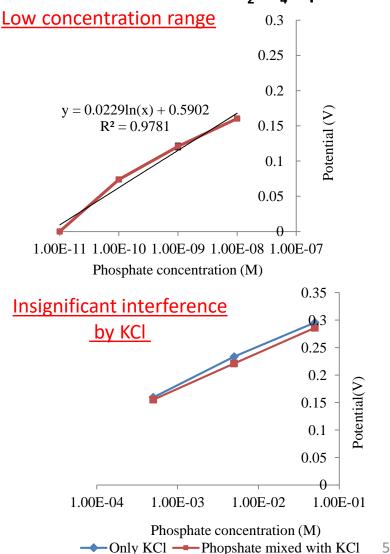
 $3CoO + 2H_2PO_4^- + 2H^+ \rightleftharpoons Co_3(PO_4)_2 + 3H_2O \text{ at pH } 4.0$  $3CoO + 2HPO_4^{2-} + H_2O \rightleftharpoons Co_3(PO_4)_2 + 4OH^- \text{ at pH } 8.0$  $3CoO + 2PO_4^{3-} + 3H_2O \rightleftharpoons Co_3(PO_4)_2 + 6OH^- \text{ at pH } 11.0$ 

Carbon working electrode surface of Screen Printed Electrode (SPE) modified by selective combinations of Graphene Oxide, Pyrrole, Cobalt Oxide nanoparticles, Tin (IV) Chloride, Diphenyltin Dichloride, or Ammonium Molybdate by drop-casting method

□ Different concentrations of  $KH_2PO_4$  aqueous solution used to determine the sensitivity and lower detection limit (LOD) of phosphate ion ( $PO_4^{3-}$ ) of the developed sensor using open circuit voltammetry

# **UWM** Phosphate Detection in Several Range

Phosphate detection using mixture of Pyrrole and Ammonium molybdate modified SPE in KH<sub>2</sub>PO<sub>4</sub> aqueous solution at pH 4.5 (left)



 High concentration range
 0.3 

  $y = 0.0067 \ln(x) + 0.2784$  0.3 

  $R^2 = 0.9864$  0.2 

 0.15 0.15 

 0.15 0.15 

 0.15 0.15 

 0.05 0.05 

 1.00E-09 1.00E-06 1.00E-03 

 Phosphate concentration (M)

Interference test of Phosphate ions in presence of Cl<sup>-</sup> ions using mixture of Pyrrole and Ammonium molybdate modified SPE in KCl solution and KCl in KH<sub>2</sub>PO<sub>4</sub> aqueous solution at pH 4.5

# WM Intellectual Property and Licensing

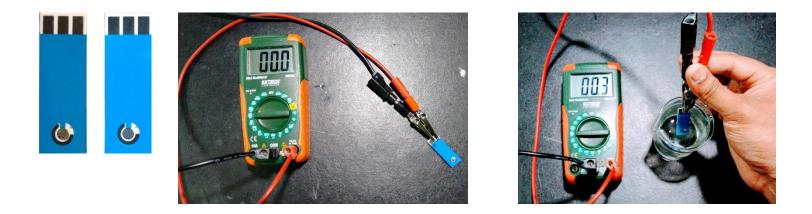
## **PCT** Filed Oct. 2018, PCT/US2018/046322

## **Current Status**

- Proposed technology is a part of active on going research program at UW-Milwaukee
- Seeking development partners for testing to aid in final end user prototype
- Technology is current available for licensing under non-exclusive terms



## **In Summary**



- Our team has demonstrated a highly sensitive phosphate sensor that can be used as a simple hand-held device
- □ The materials for manufacture are inexpensive and easy to obtain
- Extensive training will not be necessary to use the device
- □ Licensees can derive further profit from the use of one time disposable sensors



Determine whether sensors can be used in longer term applications or continuous use:

- Sensor will be submerged in standard solution for certain period of time, and then use it for the measurements to characterize the sensitivity change over time and robustness
- Test sensors further to characterize effect of temperature (5°C-50°C) on detection, as well as appropriate temperature range for the detection
  - The sensor will be tested in pressure chamber with pressure up to 80 psi (the pressure regulator in residential use is set between 40-50 psi)

□ Find partner to manufacture and develop the final prototypes



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For further information please contact:

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